# TOTAL SYNTHESIS OF (+)-ASTELTOXIN1

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Abstract: A total synthesis of (+)-asteltoxin, a novel mycotoxin isolated from <u>Aspergillus stellatus</u>, has been achieved by using D-glucose as an enantiomerically pure starting material.

In 1979 Vleggaar and co-workers reported the isolation of a novel mycotoxin (+)-asteltoxin (1) from the toxic maize meal cultures of <u>Aspergillus stellatus</u> Curzi (MRC 277). Based on a combination of the spectral analysis and a single-crystal X-ray crystallography, the relative stereochemistry of 1 was determined. A notable structural characteristic of 1 is the bistetrahydrofuran moiety having six consecutive stereogenic centers. The structural similarity of 1 to the other mycotoxins such as (-)-citreoviridin (2) and (-)-aurovertin B (3) to the potent inhibitors of ATP-synthesis and ATP-hydrolysis catalyzed by mitochondrial enzyme systems, implicated similar bioactivity for 1. In fact, asteltoxin (1) was found to be a potent inhibitor of E. Coli. BF<sub>1</sub>-ATPase activity. Biosynthetic studies of the unusual bistetrahydrofuran moiety of 1 have been disclosed by Vlaggaar. In the past several years efforts have resulted in the total syntheses of 1, 2, 3, 8 and related mycotoxins.

1 (+)-ASTELTOXIN

(-)-AUROVERTIN B

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In 1983, the first synthesis of racemic bistetrahydrofuran moiety of 1 was reported by Schreiber and Satake. Soon afterward, they completed the first total synthesis of  $(\pm)-1$ . This total synthesis confirmed the relative configuration of 1. They finally established the absolute stereochemistry of natural (+)-1 in 1986 through an asymmetric synthesis of the bistetrahydrofuran moiety. Herein, we report a total synthesis of natural (+)-1, which features a carbohydrate-mediated chiron approach. 13

As an enantiomerically pure starting compound for our total synthesis of 1, we used a derivative of tetrahydrofuran (4) which includes a stereochemically defined quaternary carbon atom. The compound 4 was prepared from D-glucose by employing the ortho ester Claisen rearrangement as a key reaction. <sup>14</sup> The quaternary carbon atom corresponds to the C-5 in 1 (asteltoxin numbering), and modification of the vinyl group in 4 would afford the C-7 through C-9 carbon framework. For this purpose, the vinyl group in 4 was cleaved by ozonolysis to afford an aldehyde (5) (Scheme 1). Wittig reaction of 5 with  $Ph_3P$ =CHCOOEt in refluxing benzene afforded the (E)- $\alpha$ . $\beta$ -unsaturated ester (6E) in 90% yield along with a 4% yield of 6Z (J=18 Hz for the vinyl protons of 6E). The predominant isomer 6E possesses the proper double bond geometry for functionalization in

**a.**  $0_3$ ,  $CH_2C1_2$ , -78 °C; then  $Ph_3P$ , -78 °C to r.t.. **b.**  $Ph_3P$ =CHCOOEt, benzene, reflux.

c. Dibal-H, CH<sub>2</sub>Cl<sub>2</sub>, -40 °C. d. BzlBr, NaH, DMF, r.t.. e. 50% aqueous AcOH, r.t..

f.  $Me_3CC(0)Cl$ , pyridine, DMAP,  $CH_2Cl_2$ , r.t.. g. 60% aqueous  $CF_3COOH$ , 5 °C, 64 h. h.  $NaIO_4$ ,  $H_2O/MeOH$ , r.t.. i. MeOH, Amberlite IR-12O ( $H^+$ ), reflux.

later stages. Diisobutylaluminum hydride (Dibal-H) reduction of 6E followed by benzylation of the allyl alcohol (7) gave the benzyl ether (8) in 94% yield. Hydrolysis of 8 with 50% AcOH afforded the diol (9) which was then selectively acylated to give the mono-pivaloyl ester (10) in 89% yield. The isopropylidene group in 10 was removed upon exposure to 60% CF $_3$ COOH at 5 °C to give a diastereomeric mixture of hemiacetal (11) in 78% yield. NaIO $_4$  oxidation of 11 for glycol cleavage afforded a pentasubstituted tetrahydrofuran (12), which was converted to the diastereomerically single methyl glycoside (13) in 80% yield by heating in MeOH in the presence of (H $^+$ )- type resin. The formyl group in 12 was removed under these glycosidation conditions. The anomeric configuration of 13 was determined based on the n.O.e. experiment of the advanced compound 17.

Introduction of a cis-diol to the double bond in 13 was next executed. (Scheme 2). The hydroxy group in 13 was protected to give the benzoate (14) quantitatively. tetroxide oxidation of 14 under the Kelly's conditions 16 [catalytic OsO, in the presence of N-methylmorpholine N-oxide (NMO)] provided the cis-diol (15) in 68% yield along with the diastereomer (16) in 15% yield. The stereochemistry of the cis-diol in the main product 15 was confirmed to be the desired  $\underline{R},\underline{R}$ -configuration by some chemical transformations. 18 The steric environment of 14 seems to be the major factor for this stereoselective cis-dihydroxylation. The hydroxyl groups in 15 were protected to be an acetonide (17), which was then exposed to MeONa briefly at 0 °C affording the debenzoyl derivative (18) ın 62% yield. The liberated hydroxyl group was then benzylated, and the pivaloyl group in the resulting benzyl ether (19) was removed with excess MeONa at room temperature to afford the primary hydroxyl derivative (20) in 83% yield. oxidation <sup>19</sup> of 20 smoothly gave an aldehyde (21), which was subjected to Wittig methylenation under the standard conditions affording the adduct (22) in 87% yield. Simultaneous saturation of the vinyl group and removal of the benzyl groups in 22 under a prolonged hydrogenation in the presence of Raney N1 resulted in the formation of 23 For functionalization of the secondary hydroxyl group in 23. (78%). hydroxyl group was protected to be the pivaloyl ester (24) in 87% yield. chlorochromate (PCC) oxidation of 24 gave a tetrahydrofuranone derivative (25). Grignard addition of MeMgBr to 25 in THF at 0 °C proceeded highly stereoselectively, to our delight, to afford one diastereomer (26) in 95% yield. The homogeneity of the adduct was confirmed by the <sup>1</sup>H NMR (400 MHz) analysis. The stereochemistry of the newly introduced stereogenic center was establised to be (S) as that of natural 1 through the transformation of 26 to the bistetrahydrofuran moiety of 1 , Schreibers's key intermediate for their total synthesis, by acid-catalyzed cyclization.<sup>21</sup> account for this exclusively stereoselective addition of the Grignard reagent is that the reagent is directed to the  $\beta$ -face of the tetrahydrofuranone ring by chelation with one of the isopropylidene ketal oxygens. As a result, the attack of the reagent took place from the  $\beta$ -face.

The final stage of the total synthesis was effected as follows (Scheme 3). PCC oxidation of 26 followed by Wittig-Horner reaction of the resulting aldehyde (27) with triethyl 4-phosphonocrotonate provided  $(E,E)-\alpha\beta,\gamma\delta$ -unsaturated ester (28) in 72% yield. Acid hydrolysis (60% aqueous CF<sub>3</sub>COOH) of 28 smoothly gave a bistetrahydrofuran (29) in 89% yield. Dibal-H reduction of 29 provided the allyl alcohol (30) in 78% yield.

a. BzCl, pyridine, 60 °C. b. 0sO $_4$  in 2-methyl-2-propanol, NMO, aqueous acetone, r.t.. c. 2,2-dimethoxypropane, acetone, CSA, r.t.. d. MeONa, MeOH, 0 °C, 90 min. e. NaH, BzlBr, DMF, r.t.. f. MeONa, MeOH, r.t., 20 h. g. DMSO,  $(COCl)_2$ ,  $CH_2Cl_2$ , - 78 °C; then Et $_3$ N, -78 °C to r.t.. h.  $Ph_3P^+MeBr^-$ ,  $NaNH_2$ , THF, reflux; then added 21. THF, r.t.. i.  $H_2$ , Raney Ni, EtOH, r.t., 4 days. j.  $Me_3CC(0)Cl$ , pyridine, DMAP,  $CH_2Cl_2$ , r.t.. k. PCC, MS,  $CH_2Cl_2$ , r.t. 1. MeMgBr, THF, 0 °C.

## Scheme 2

Active MnO<sub>2</sub> oxidation of 30 gave  $(E,E)-\alpha\beta$ ,  $\gamma\delta$ -unsaturated aldehyde (31), an advanced intermediate for the Schreiber's total synthesis of 1, in 86% yield. The aldol condensation protocol previously reported was used for our final two steps. The aldol reaction of 31 and 4-methoxy-5,6-dimethyl-2-pyrone (32)<sup>22</sup> with LDA (lithium diisopropylamide) provided a diastereomeric mixture (33) in 28% yield (the aldehyde 32 was recovered in 32% yield, however, this reaction was not optimized). By treatment of this mixture 33 with excess p-toluenesulfonyl chloride in the presence of Et<sub>3</sub>N and 4-dimethylaminopyridine (DMAP) gave (+)-asteltoxin (1) in 73% yield. The  $^1\text{H}$  (400 MHz) and  $^{13}\text{C}$  (100 MHz) NMR, and MS of the synthetic 1 were superimposable with those of natural and racemic ones. The TLC behavior in several solvent systems revealed that the synthetic 1 was identical with natural source. The specific rotation of 1 [[ $\alpha$ ]<sub>D</sub> +20° (c 0.1, MeOH)] well matched with that for natural product [[ $\alpha$ ]<sub>D</sub>+20° (c 1.15, MeOH)].

In conclusion, we completed the total synthesis of (+)-1. Through the present synthesis, the utility of the D-glucose derived synthon 4 was embodied.

a. PCC, MS,  $\text{CH}_2\text{Cl}_2$ , r.t.. b.  $(\text{Et0})_2\text{P}(0)\text{CH}_2\text{CH}=\text{CHCOOEt}$ , LDA, THF, -78 °C; then added 27, -78 °C to r.t.. c. 60% aqueous  $\text{CF}_3\text{COOH}$ , r.t.. d. Dibal-H,  $\text{CH}_2\text{Cl}_2$ , -78 °C. e.  $\text{MnO}_2$ ,  $\text{CH}_2\text{Cl}_2$ , r.t.. f. LDA, THF, HMPA, -78 °C, added 32; then added 31, -78 °C. 15 min. g. TsCl, DMAP,  $\text{Et}_3\text{N}$ ,  $\text{CH}_2\text{Cl}_2$ , r.t..

## Scheme 3

### **EXPERIMENTAL**

Melting points are uncorrected. Specific rotations in CHCl, were measured by using JASCO Model DIP-4 polarimeter in a 10 mm cell. IR spectra (neat) were recorded on a JASCO Model A-202 spectrometer.  $^1\mathrm{H}$  NMR spectra were recorded on a Varian EM-390 (90 MHz) and JEOL JNM GX-400 (400 MHz) spectrometers in CDCl $_3$  solution with tetramethylsilane as an internal standard.  $^{13}\mathrm{C}$  NMR spectra were obtained on a JEOL JNM GX-400 (100 MHz) spectrometer. High resolution mass spectra (HRMS) were measured by using a Hitachi M-80 spectrometer (EI method). Thin-layer chromatography (TLC) was performed with a glass plate coated with Kieselgel 60 GF $_{254}$  (Merck). Each crude product was specified, reactions were carried out at room temperature. Reactions involving organometallic reagents were performed under argon atmosphere. Solvents were removed by concentration using an evaporator.

(2R,3R,4R,5S)-4-[(E)-2-(Ethoxycarbonyl)ethenyl]-2,3-(isopropylidenedioxy)-5-[(1R)-1.2-(isopropylidenedioxy)ethyl]-4-methyltetrahydrofuran (6E). To a solution of 4 (5.02 g, 17.7 mmol) in  $\mathrm{CH_2Cl_2}$  (40 mL) was bubbled a stream of  $\mathrm{O_3}$  (ca. 3% in  $\mathrm{O_2}$ ) at -78 °C until blue color of the solution was sustained. A  $\mathrm{CH_2Cl_2}$  solution (40 mL) of Ph P (9.27 g, 35.5 mmol) was added to the mixture, and gradually warmed to room temparature. After stirring for 1 h, the mixture was concentrated. The residue was rapidly chromatographed on silica gel (AcOEt/hexane 1:15) to give an aldehyde 5 (5.16 g), which was subjected to the Wittig reaction directly: TLC R<sub>f</sub> 0.40 (AcOEt/hexane 1:3); <sup>1</sup>H NMR (90 MHz) &1.12 (3H, s), 1.30, 1.31, 1.38, 1.59 (3H x 4, 4s), 4.09 (2H, d, J=3 Hz), 4.10 (1H, ddd, J=8, 3, 3 Hz), 4.40 (1H, d, J=4 Hz), 4.54 (1H, d, J=8 Hz), 5.84 (1H, d, J=4 Hz), 9.74 (1H, s).

A solution of 5 (5.16 g) and Ph<sub>3</sub>P=CHCOOEt (24.6 g, 70.6 mmol) in benzene (100 mL) was refluxed for 12 h. The mixture was concentrated in vacuo, and the residue was stirred with petroleum ether. After cooling at 5 °C overnight, the precipitated Ph<sub>3</sub>PO was removed by filtration, washed well with petroleum ether. The combined filtrate and washings were concentrated. The residue was chromatographed on silica gel (AcOEthexane 1:20, 1:10, then 1:5) to give 6Z (239 mg, 4%) and 6E (5.64 g, 90%). 6Z as a colorless oil: TLC R<sub>E</sub> 0.55 (AcOEthexane 1:3): [ $\alpha$ ] $_{2}^{6}$ +108.9° (c 1.63); IR 3000, 2950, 2880, 1725, 1645, 1460, 1420, 1380, 1375 cm<sup>-1</sup>: 1H NMR (90 MHz) $\delta$ 1.3-1.5 (18H. m), 3.9-4.3 (6H, m), 5.01 (1H, d, J=5 Hz), 5.72 (1H, d, J=4 Hz), 5.85 (1H, d, J=13 Hz), 6.35 (1H, d, J=13 Hz); HRMS calcd for C<sub>17</sub>H<sub>25</sub>O<sub>7</sub>: m/z 341.1597 (M<sup>+</sup>-CH<sub>3</sub>). Found: m/z 341.1602. 6E as a colorless oil: TLC R<sub>E</sub> 0.46 (AcOEt/hexane 1:3): [ $\alpha$ ] $_{2}^{6}$ 5+93.0° (c1.11); IR 2980, 2930, 2870, 1715, 1655, 1450, 1370, 1300 cm<sup>-1</sup>; 1H NMR (90 MHz) $\delta$ 1.1-1.6 (18H, m), 3.9-4.3 (7H, m), 5.79 (1H, d, J=4 Hz), 5.94 (1H, d, J=18 Hz), 7.12 (1H, d, J=18 Hz); HRMS calcd for C<sub>17</sub>H<sub>25</sub>O<sub>5</sub>: m/z 341.1597 (M<sup>+</sup>-CH<sub>3</sub>). Found: m/z 341.1593.

(2R, 3R, 4R, 5S)-4-[(E)-3-(Benzyloxy)-1-propenyl]-2,3-(isopropylidenedioxy)-5-[(1R)-1,2-(isopropylidenedioxy)ethyl]-4-methyltetrahydrofuran (8). To a stirred solution of 6E (5.45 g, 15.3 mmol) in  $\text{CH}_2\text{Cl}_2$  (100 mL) was added Dibal-H (25 wt% in toluene, 30.9 mL, 45.9 mmol) at -40 °C. After stirring at -40 °C for 30 min, the mixture was quenched with 10 mL of  $\text{H}_2\text{O}$ . After warming to room temperature, the resulting gels were removed by filtration, washed well with  $\text{CH}_2\text{Cl}_2$ . The combined filtrate and washings were washed with  $\text{H}_2\text{O}$  (150 mL). The aqueous phase was extracted with  $\text{CH}_2\text{Cl}_2$  (100 mL x 4). The combined organic phases were dried ( $\text{Na}_2\text{SO}_4$ ) and concentrated in vacuo to give 7 (4.90 g), which was benzylated directly. In a separate experiment, the crude 7 was purified by silica gel chromatography (AcOEt/hexane 1:3): TLC  $\text{R}_f$  0.20 (AcOEt/hexane 1:2):  $[\alpha]_D^{26}$  +67.1° (c 0.98); IR 3470, 2980, 2940, 2880, 1450, 1370, 1245, 1215 cm<sup>-1</sup>; <sup>1</sup>H NMR (90 MHz) 1.02 (3H, s), 1.26, 1.38, 1.48 (6H, 3H, 3H, 3s), 2.0-2.2 (1H, br s), 3.9-4.1 (7H, m), 5.6-5.8 (3H, m); HRMS calcd for  $\text{C}_1\text{GH}_2\text{SO}_6$ : m/z 313.1649 (M+-H). Found: m/z 313.1648.

To a suspension of NaH (734 mg, 30.6 mmol) in DMF (20 mL) were added a DMF solution (40 mL) of 7 (4.90 g) and benzyl bromide (3.6 mL, 30.6 mmol) at 0 °C. After stirring for 4 h, 5 mL of EtOH was added to the mixture. The mixture was stirred for 30 min and then concentrated in vacuo. The residue was partitioned between AcOEt (300 mL) and  $\rm H_2O$  (100 mL). The aqueous phase was extracted with AcOEt (100 mL x 2). The combined organic phases were dried ( $\rm Na_2SO_4$ ) and concentrated. The residue was chromatographed

on silica gel (AcOEt/hexane 1:15) to give 8 (5.80 g, 94%) as a colorless oil: TLC  $R_{\rm f}$  0.74 (AcOEt/hexane 1:2);  $[\alpha]_{\rm c}^{26}$  +47.5° (c 1.04); IR 2980, 2930, 2870, 1490, 1450, 1375, 1365, 1210 cm $^{-1}$ ;  $^{1}$  H NMR (90 MHz)  $\delta$  1.09 (3H, s), 1.30, 1.35, 1.54 (6H, 3H, 3H, 3s), 4.0–4.2 (7H, m), 4.53 (2H, s), 5.78 (1H, d, J=4.5 Hz), 5.8–6.1 (2H, m), 7.36 (5H, s); HRMS calcd for  $C_{23}H_{32}O_6$ : m/z 404.2196 (M+). Found: m/z 404.2176.

To a solution of 9 obtained above in CH<sub>2</sub>Cl<sub>2</sub> (150 mL) were added pyridine (50 mL), pivaloyl chloride (3.5 mL, 28.7 mmol), and DMAP (526 mg, 4.3 mmol). The mixture was stirred for 10 h, during which 2 mL of Et<sub>3</sub>N and each 1.8 mL of pivaloyl chloride were added after 5 and 8 h. The mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub> (100 mL), washed with H<sub>2</sub>O (200 mL x 1, 100 mL x 1). The organic phase was dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated. The residue was chromatographed on silica gel (AcOEt/hexane 1:10 then 1:5) to give 10 (5.71 g, 89%) as a colorless oil: TLC R<sub>f</sub> 0.65 (AcOEt/hexane 1:2); [ $\alpha$ ]<sup>25</sup> +62.2° (c 1.31); IR 3500, 2980, 2940, 2870, 1710, 1480, 1460 cm<sup>-1</sup>; <sup>1</sup>H NMR (90 MHz)  $\delta$  1.08 (3H, s), 1.18 (9H, s), 1.24, 1.48 (3H x 2, 2s), 2.03 (1H, br s), 4.0-4.3 (7H, m), 4.46 (2H, s), 5.7-6.1 (3H, m), 7.30 (5H, s); LRMS m/z 433 (M<sup>+</sup> -CH<sub>3</sub>).

(2R,3R,4S,5R)-3-[(B)-3-(Benzyloxy)-1-propenyl]-5-[[(2,2-dimethylpropionyl)oxy] methyl]-4-hydroxy-2-methoxy-3-methyltetrahydrofuran (13). A solution of 10 (5.71 g, 12.7 mmol) in 60% aqueous  $CF_3COOH$  (83 mL) was stirred at 5 °C for 64 h. The solution was neutralized with 10 M aqueous NaOH (ca. 60 mL) and diluted with  $H_2O$  (100 mL). This was extrated with  $CH_2Cl_2$  (200 mL x 2, 100 mL x 4). The combined organic phases were dried ( $Na_2SO_4$ ) and concentrated. The residue was chromatographed on silica gel (AcOEt/hexane 1:2 then 2:3) to give an anomeric mixture of 11 (4.07 g, 78%) as a colorless oil: IR 3430, 2970, 2930, 2910, 2870, 1720, 1480, 1450 cm<sup>-1</sup>:  $^{1}$ H NMR (90 MHz)& 1.21 (9H, s), 1.30 (3H, s), 2.2-2.6 (3H, m), 3.0-4.7 (7H, m), 4.53 (2H, s), 5.1-5.2 (1H, m), 5.8-6.2 (2H, m), 7.36 (5H, s).

To a stirred solution of 11 (4.07 g, 9.9 mmol) in MeOH (200 mL) was added an aqueous solution (80 mL) of NaIO<sub>4</sub> (6.37 g, 29.8 mmol). The mixture was stirred for 2 h, and the precipitated solids were removed by filtration , washed well with MeOH. The combined filtrate and washings were concentrated in vacuo. The residue was dissolved in  $\rm H_2O$  (200 mL), and extracted with  $\rm CH_2Cl_2$  (200 mL x 2, 100 mL x 3). The combined organic phases were dried ( $\rm Na_2SO_4$ ) and concentrated in vacuo to give crude 12 which was used directly. In a separate experiment, the crude 12 was purified by silica gel chromatography. 12 as a colorless oil: TLC Rf 0.75 (AcOEt/hexane 2:3); IR 3450, 2980, 2930, 2870, 1730, 1480, 1455, 1360, 1280, 1160 cm<sup>-1</sup>;  $^1\rm H$  NMR (90 MHz)  $^5\rm G$  1.22 (9H, s), 1.25 (3H, s), 3.6 (1H, br s), 4.0-4.1 (2H, m), 4.2-4.3 (3H, m), 4.51 (2H, s), 5.0-5.4 (2H, m), 5.7-6.1 (2H, m), 7.33 (5H, s), 8.02 (1H, s).

A solution of 12 obtained above in MeOH (80 mL) was heated at 60 °C for 7 h in the presence of Amberlite IR-120 (H<sup>+</sup>) (ca. 6 g). The resin was removed by filtration and washed well with MeOH. The combined filtrate and washings were concentrated in vacuo. The residue was chromatographed on silica gel (AcOEt/hexane 1:10) to give 13 (2.59 g, 67%) as a colorless oil. The unreacted 12 (705 mg, 17%) was also recovered, which was recyclized for the acetalization to give 13 (520 mg). Total amount of 13 was 3.11 g (80 %) after one recycle. 13: TLC R<sub>f</sub> 0.46 (AcOEt/hexane 1:3); [ $\alpha$ ]<sup>26</sup> -0.5°(c 0.80); IR 3450, 2970, 2940, 1730, 1720, 1480, 1285, 1160 cm<sup>-1</sup>; <sup>1</sup>H NMR (90 MHz)<sup>D</sup>6 1.17 (3H, s), 1.22 (9H, s), 2.0 (1H, br s), 3.33 (3H, s), 3.8-4.3 (6H, m), 4.51 (2H, s), 4.56 (1H, s), 5.8-5.9 (2H, m), 7.33 (5H, s); HRMS calcd for C<sub>22</sub>H<sub>32</sub>O<sub>6</sub>: m/z 392.2196 (M<sup>+</sup>). Found: m/z 392.2187.

 $(2R,3S,4S,5R)-3-[(1R,2R)-[3-(Benzyloxy)-1,2-(isopropylidenedioxy)propyl]-5-[[(2,2-dimethylpropionyl)oxy]methyl]-4-hydroxy-2-methoxy-3-methyltetrahydrofuran (18). A solution of 15 (1.06 g, 2.0 mmol) in a mixture of acetone (40 mL) and 2,2-dimethoxypropane (1.2 mL, 10.0 mmol) was stirred for 42 h in the presence of camphorsulfonic acid (87 mg). The solution was neutralized with saturated aqueous NaHCO3 and concentrated in vacuo. The residue was partitioned between CH2Cl2 (200 mL) and H2O (200 mL), and the aqueous phase was extracted with CH2Cl2 (100 mL x 2). The combined organic phases were dried (Na2SO4) and concentrated to give crude 17 (1.17 g), which was used for the next step directly. In a separate experiment, the crude 17 was purified by silica gel chromatography. 17 as a colorless oil: TLC Rf 0.66 (AcOEt/hexane 1:3); [<math>\alpha$ ] $_D^{2B}$  -10.0° (c 1.21); IR 2990, 2950, 1740, 1600, 1480, 1460, 1400, 1380, 1370, 1320, 1270 cm-1;  $\alpha$ 1 h NMR (400 MHz)  $\alpha$ 2 1.06 (3H, s), 1.18 (9H, s), 1.39, 1.48 (3H x 2, 2s), 3.41 (3H, s), 3.49-3.61, 4.09-4.47 (2H, 5H, 2m), 4.49, 4.54 (1H x 2, 2d, J=12.4 Hz), 4.94 (1H, s), 5.44 (1H, d, J=6.8 Hz), 7.26-7.60, 8.03-8.06 (10H, m); HRMS calcd for C31H39O9; m/z 555.2583.

To a solution of 17 (1.17 g) in MeOH (40 mL) was added MeONa (1 M solution in MeOH, 0.8 mL, 0.8 mmol) at 0 °C. The mixture was stirred for 90 min, and neutralized with Amberlite IR-120 (H<sup>+</sup>). The resin was removed by filtration, washed well with MeOH. The combined filtrate and washings were concentrated in vacuo. The residue was chromatographed on silica gel (AcOEt/hexane 1:7) to give 18 (577 mg, 62%) as a colorless oil: TLC R, 0.34 (AcOEt/hexane 1:3);  $\left[\alpha\right]_{30}^{30}$  -24.2 °(c 1.23); IR 3430, 2980, 2960, 2870, 1730, 1480, 1460, 1380, 1360, 1280 cm<sup>-1</sup>; <sup>1</sup>H NMR (90 MHz)  $\delta$  1.06 (3H, s), 1.22 (9H, s), 1.36, 1.42 (3H x 2, 2s), 3.2 (1H, br s), 3.32 (3H, s), 3.5-4.4 (8H, m), 4.58 (2H, s), 4.80

(1H, s), 7.32 (5H, s); HRMS calcd for  $\mathrm{C_{24}H_{35}O_{8}}$ : m/z 451.2329 (M+-CH<sub>3</sub>). Found: m/z 451.2325.

(2R, 3S, 4S, 5R)-4-(Benzyloxy)-3-[(1R, 2R)-[3-(benzyloxy)-1,2-(isopropylidenedioxy)-propyl]-5-(hydroxymethyl)-2-methoxy-3-methyltetrahydrofuran (20). To a solution of 18 (553 mg, 1.2 mmol) in DMF (20 mL) were added dried NaH (57 mg, 2.4 mmol) and benzyl bromide (0.4 mL, 3.6 mmol). The mixture was stirred for 15 h and 0.2 mL of MeOH was added. The mixture was diluted with AcOEt (200 mL) and washed with  $\rm H_2O$  (50 mL x 3). The organic phase was dried (Na $_2$ SO $_4$ ) and concentrated in vacuo to give crude 19 which was deacylated directly. In a separate experiment, the crude 19 was purified by PTLC to give pure 19: TLC R $_f$  0.57 (AcOEt/hexane 1:4); [ $\alpha$ ] $_0^{30}$  +6.3°(c 1.16); IR 2990, 2960, 2940, 2870, 1730, 1500, 1460, 1400, 1380, 1360, 1280, 1250, 1170 cm<sup>-1</sup>; <sup>1</sup>H NMR (90 MHz)  $\delta$  1.02 (3H, s), 1.19 (9H, s), 1.31, 1.37 (3H x 2, 2s), 3.24 (3H, s), 3.4-4.4 (12H, m), 4.67 (1H, s), 7.23 (10H,s); HRMS calcd for  $C_{31}H_{41}O_8$ : m/z 541.2798 (M+-CH $_3$ ). Found: m/z 541.2788.

To a solution of 19 obtained above in MeOH (15 mL) was added MeONa (1 M solution in MeOH, 5.9 mL, 5.9 mmol). The mixture was stirred for 20 h, and neutralized with Amberlite IR-120 (H $^+$ ). The resin was removed by filtration, washed well with MeOH. The combined filtrate and washings were concentrated. The residue was chromatographed on silica gel to give 20 (465 mg, 83%) as a colorless oil: TLC  $\rm R_f$  0.45 (AcOEt/hexane 1:2);  $\rm [\alpha]_{31}^{31}$  +8.6° (c 0.80); IR 3480, 2990, 2910, 2880, 1500, 1450, 1380, 1370, 1260 cm $^{-1}$ ;  $\rm ^{1}H$  NMR (90 MHz)  $\delta$  1.05 (3H, s), 1.37, 1.43 (3H x 2, 2s), 2.0 (1H, br s), 3.38 (3H, s), 3.4-4.5 (12H, m), 4.75 (1H, s), 7.30 (10H, s); HRMS calcd for  $\rm C_{27}H_{37}O_{7}$ : m/z 473.2537 (M++H). Found: m/z 473.2511.

To a solution of 21 obtained above in benzene (20 mL) was added Ph\_P=CH $_2$  [754 mg (2.7 mmol), prepared from Ph $_3$ PCH $_3$ Br by treatment with fresh NaNH $_2$  in refluxing THF for 4 h followed by concentration of the supernatant part of the reaction mixture]. The mixture was stirred for 10 min and H $_2$ O (100 mL) was added, then extracted with Et $_2$ O (100 mL x 1, 50 mL x 2). The combined organic phases were dried (Na $_2$ SO $_4$ ) and concentrated in vacuo. The residue was chromatographed on silica gel (AcOEt/hexane 1:30) to give 22 (370 mg, 87%) as a colorless oil: TLC R $_f$  0.67 (AcOEt/hexane 1:4); [ $\alpha$ ] $^{25}$  -19.2° (c 1.00); IR 2990, 2940, 2910, 2870, 2830, 1640, 1500, 1450, 1390, 1370, 1250, 1210 cm $^{-1}$ ;  $^{11}$ H NMR (90 MHz)  $^{5}$  1.07 (3H, s), 1.37, 1.45 (3H x 2, 2s), 3.38 (3H, s), 3.5-4.5 (10H, m), 4.78 (1H, s), 5.1-5.4 (2H, m), 5.7-6.0 (1H, m), 7.30 (10H, s); HRMS calcd for  $^{6}$ C $_2$ 7H $_3$ 3O $_6$ : m/z 453.2274 (M+-CH $_3$ ). Found: m/z 453.2273.

(2R,3S,4S,5R)-5-Ethyl-4-hydroxy-3-[(1R,2R)-[3-hydroxy-1,2-(isopropylidenedioxy)]-propyl]-2-methoxy-3-methyltetrahydrofuran (23). A solution of 22 (370 mg, 0.8 mmol) in EtOH (5 mL) was stirred for 4 days under atmospheric hydrogen in the presence of Raney nickel. The catalyst was removed by filtration through a Celite-pad, washed well with EtOH. The combined filtrate and washings were concentrated in vacuo. The residue was chromatographed on silica gel (AcOEt/hexane 1:4 then 1:2) to give 23 (178 mg, 78%) as colorless needles, mp 59.5-61 °C: TLC R<sub>f</sub> 0.20 (AcOEt/hexane 1:2); [ $\alpha$ ]<sup>29</sup> -31.1 (c 1.10); IR 3400, 2990, 2970, 2940, 2890, 2840, 1460, 1380, 1370, 1250, 1210 cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz)  $\delta$  1.02 (3H, t, J=7.3 Hz), 1.06 (3H, s), 1.40 , 1.45 (3H x 2, 2s), 1.57-1.74 (2H, m), 2.32 (1H, br s), 2.90 (1H, br s), 3.42 (3H, s), 3.53-3.58 (1H, m), 3.66-3.74 (2H,

- m), 3.84 (1H, d, J=11.7 Hz), 4.13-4.19 (2H, m), 4.86 (1H, s). Anal. Calcd for  $C_{14}H_{26}O_6$ : C, 57.91; H, 9.03. Found: C, 58.27; H, 9.00.

was dried (Na $_2$ SO $_4$ ) and concentrated in vacuo. The residue was purified by PTLC (AcOEt/hexane 1:4) to give 28 (16.6 mg, 72%) as a colorless oil: TLC R $_f$  0.60 (AcOEt/hexane 1:2); [ $\alpha$ ] $_0^2$  +55.4°(c 0.83); IR 3520, 2980, 2940, 2880, 1720, 1645, 1460, 1450, 1380, 1370, 1310, 1260, 1230 cm $^{-1}$ ; <sup>1</sup>H NMR (400 MHz)  $\delta$  0.92 (3H, s), 1.05 (3H, t, J=7.3 Hz), 1.12 (3H, s), 1.30 (3H, t, J=7.3 Hz), 1.36, 1.45 (3H x 2, 2s), 1.53–1.60 (2H, m), 2.74 (1H, s), 3.39 (3H, s), 3.83 (1H, d, J=6.8 Hz), 3.93 (1H, dd, J=4.9, 8.3 Hz), 4.21 (2H, q, J=7.3 Hz), 4.41 (1H, t, J=6.8 Hz), 4.80 (1H, s), 5.93 (1H, d, J=15.1 Hz), 6.11 (1H, dd, J=6.8, 15.1 Hz), 6.45 (1H, dd, J=10.7, 15.1 Hz), 7.28 (1H, dd, J=10.7, 15.1 Hz); HRMS calcd for C $_2$ 0H $_3$ 10 $_7$ : m/z 383.2068 (M+-CH $_3$ ). Found: m/z 383.2095.

(1R, 3R, 4R, 5R, 6R, 7R) – 3-Ethyl – 4, 6-dihydroxy-7-[(1E, 3E) – 5-hydroxypenta – 1, 3-dienyl] – 4, 5-dimethyl – 2, 8-dioxabicyclo[3.3.0] octane (30). To a solution of 29 (11.0 mg, 0.03 mmol) in CH $_2$ Cl $_2$ (1.5 mL) was added Dibal-H (25 wt% in toluene, 0.23 mL, 0.03 mmol) at –78 °C. The mixture was stirred at –78 °C for 2 h during which 0.23 mL of Dibal-H was added after 1 h. The mixture was quenched with H $_2$ O (0.1 mL), and warmed to room temperature. Then, 5 mL of THF and 2 g of Na $_2$ SO $_4$  were added. The inorganic solids were removed by filtration, and washed well with AcOEt. The combined filtrate and washings were concentrated in vacuo. The residue was purified by PTLC (EtOH/toluene 1:6) to give 30 (7.5 mg, 78%) as a colorless oil: TLC R $_f$  0.25 (EtOH/toluene 1:5); [ $\alpha$ ] $_5^2$ 3 +30.0° (c 0.38); IR 3400, 2970, 2930, 2880, 2860, 1660, 1630, 1460, 1380, 1355, 1290, 1260, 1200 cm $^{-1}$ ; <sup>1</sup>H NMR (400 MHz)  $\delta$  1.06 (3H, t, J=7.3 Hz), 1.18 (3H, s), 1.38 (3H, s), 1.52–1.60 (4H, m), 1.75 (1H, d, J=4.9 Hz), 3.69 (1H, dd, J=2.9, 4.9 Hz), 4.20–4.23 (2H, m), 4.29 (1H, dd, J=4.9, 7.3 Hz), 4.68–4.72 (1H, m), 5.27 (1H, s), 5.72 (1H, dd, J=5.4, 15.1 Hz); 5.91–5.97 (1H, m), 6.32 (1H, dd, J=10.7, 15.1 Hz), 6.56 (1H, dd, J=10.7, 15.1 Hz); HRMS calcd for C $_{1.5}$ H $_{2.0}$ 0; m/z 266.1517 (M\*-H $_2$ 0). Found: m/z 266.1521.

Mixture of (1R,3R,4R,5R,6R,7R)-3-Ethyl-4,6-dihydroxy-7-[(1E,3E,5R and S)-5-hydroxy-6-(4-methoxy-5-methyl-2-pyron-6-yl)hexa-1,3-dienyl]-4,5-dimethyl-2,8-dioxabicyclo[3.3.0] octane (33). To a solution of diisopropylamine (0.04 mL, 0.26 mmol) in THF (0.8 mL) was added n-BuLi (1.64 M solution in hexane, 0.13 mL, 0.22 mmol). After the solution was stirred at 0 °C for 10 min, HMPA (0.05 mL, 0.26 mmol) was added. The solution was stirred at 0 °C for 10 min, and cooled to -78 °C. To this was added a THF solution (0.5 mL) of 32 (33.9 mg, 0.22 mmol), and the resulting yellow solution was stirred at

-78 °C for 5 min. A THF solution (2 mL) of 31 (6.2 mg, 0.02 mmol) was then added. After the mixture was stirred at -78 °C for 15 min, saturated aqueous NH4Cl (2 mL) was added. The mixture was warmed to room temperature, and diluted with H2O (10 mL). This was extracted with AcOEt (10 mL x 6). The combined organic phases were dried (Na2SO4) and concentrated. The residue was purified by PTLC (EtOH/toluene 1:3) to give 33 (2.7 mg, 28%) as a colorless oil. The aldehyde 31 (2.0 mg, 32%) was recovered. 33: TLC Rf 0.45 (EtOH/toluene 1:4); IR 3300, 2950, 2930, 2880, 2850, 1705, 1645, 1565, 1460, 1405, 1300, 1245, 1200 cm $^{-1}$ ;  $^{1}$ H NMR (400 MHz)  $\delta$  1.06 (3H, t, J=7.3 Hz), 1.18 (3H, s), 1.38 (3H, s), 1.52–1.60 (3H, m), 1.71–1.78 (3H, m), 1.91 (3H, s), 2.80–2.87 (1H, m), 3.69–3.71 (1H, m), 3.83 (3H, s), 4.30 (1H, dd, J=5.2, 7.6 Hz), 4.64–4.69 (2H, m), 5.27 (1H, s), 5.46 (1H, s), 5.72–5.84 (2H, m), 6.27–6.35 (1H, m), 6.50 (1H, dd, J=11.0, 15.9 Hz); HRMS calcd for  $C_{23}H_{30}O_7$ : m/z 418.1990 (M+ $-H_2O$ ). Found: m/z 418.1991.

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- 18. The major diastereomer 15 was converted to the bistetrahydrofuran derivative in 70% yield by exposure to 60% aqueous CF<sub>3</sub>COOH at 5 °C for 4 days. Exhaustive deacylation (MeONa, r.t.) of this product followed by isopropylidenation resulted in the formation of the isopropylidene ketal in 87% yield. The spectral evidence that the ketal was formed between two secondary hydroxyl groups verified the structures of 15 and the bistetrahydrofuran derivative.
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